



## Graphical abstract

Paper-1	Heterocyclic Letters 12: iss.-4 (2022), 697-708
<p><b>Design and synthesis of three ester-harmol derivatives: theoretical evaluation of their interaction with b1-cannabinoid receptor</b></p> <p>López-Ramos Maria, Figueroa-Valverde Lauro*, Díaz-Cedillo Francisco, Rosas-Nexticapa Marcela, Alvarez-Ramirez Magdalena, Mateu-Armad Maria Virginia, Cervantes-Ortega Catalina, Melgarejo-Guutierrez Montserrat, Priego-Delgado K.</p> <p>Laboratory of Pharmaco-Chemistry at the Faculty of Chemical Biological Sciences from the University Autonomous of Campeche,</p> <p>In this study is reported a straightforward route for synthesis of three ester-harmol derivatives. The structure of the compounds obtained was confirmed by elemental analysis, spectroscopy, and spectrometry data.</p>	
<p>The diagram illustrates the synthesis of three ester-harmol derivatives. A central harmol derivative (a benzene ring with a 2-methyl-5-nitrophenyl group and a 2-(4-chlorophenyl)amino group) is shown at the bottom. Three arrows labeled '3 stages' point upwards to three different ester-harmol derivatives: 1) a benzyl ester, 2) a phenyl ester, and 3) a 2-nitrophenyl ester.</p>	

Paper-2	Heterocyclic Letters 12: iss.-4 (2022), 709-713
<p><b>Synthesis of 8-(3-bromo-4,5-dimethoxy phenyl)-2-((4-chloro phenyl) amino)-4-imino-6-(4-nitrophenyl)-4h-pyrido [1,2-a] pyrimidine-3,9-dicarbonitrile</b></p> <p>Anil Chidrawar</p> <p>Research Center of Chemistry, Degloor College, Degloor. Dist: Nanded - 431717              S.R.T.M. University, Nanded, Maharashtra, India.              Email : <a href="mailto:anilchidrawar74@gmail.com">anilchidrawar74@gmail.com</a></p> <p>A mixture of 8-(3-bromo-4,5-dimethoxyphenyl)-6-(nitrophenyl)-4-imino-2-(methylsulfanyl)-4H-pyrido [1,2-a] pyrimidine-3,9-dicarbonitrile refluxed with 4-Chloroaniline was refluxed in presence of K<sub>2</sub>CO<sub>3</sub> in DMF as solvent for 6 to 7 hours to get 8-(3-bromo-4,5-dimethoxyphenyl)-2-((4-chlorophenyl) amino)-4-imino-6-(4-nitrophenyl)-4H-pyrido [1,2-a] pyrimidine-3,9-dicarbonitrile. The structures for the synthesized compounds are assigned on the basis of IR, <sup>1</sup>HNMR and Mass spectral studies.</p>	



Paper-3	Heterocyclic Letters 12: iss.-4 (2022), 715-718
<b>Ester Enolate-Imine Condensation to Beta Lactam: Mechanism of the Reaction</b>	
<b>Ram Naresh Yadav,<sup>1</sup> and Bimal Krishna Banik*<sup>2</sup></b>	
<sup>1</sup> Department of Chemistry, Faculty of Engineering & Technology, Veer Bahadur Singh Purvanchal University, Jaunpur-222003 (U.P.) India, <sup>2</sup> Department of Mathematics and Natural Sciences, College of Sciences and Human Studies, Deanship of Research, Prince Mohammad Bin Fahd University, Al Khobar 31952, KSA; Email: <a href="mailto:bimalbanik10@gmail.com">bimalbanik10@gmail.com</a> ; <a href="mailto:bbanik@pmu.edu.sa">bbanik@pmu.edu.sa</a>	
An ester enolate condensation with an imine produces substituted $\beta$ -lactam. The mechanism of the process is advanced to explain the stereoselectivity.	

Paper-4	Heterocyclic Letters 12: iss.-4 (2022), 719-720
<b>Banik's Rule: Rapid Heating Forms Trans Beta Lactams with Diarylimines</b>	
<b>Bimal Krishna Banik</b>	
Department of Mathematics and Natural Sciences, College of Sciences and Human Studies, Deanship of Research, Prince Mohammad Bin Fahd University, Al Khobar 31952, KSA; Email: <a href="mailto:bimalbanik10@gmail.com">bimalbanik10@gmail.com</a> ; <a href="mailto:bbanik@pmu.edu.sa">bbanik@pmu.edu.sa</a>	
Banik's Rule. The author has found rapid heating of the reactants form trans $\beta$ -lactams with diarylimines irrespective of the nature of the acid chlorides.	

Paper-5	Heterocyclic Letters 12: iss.-4 (2022), 721-723
<b>The Specificity of Bismuth Nitrate and Ceric Ammonium Nitrate in the Reaction with N-Anisyl <math>\beta</math>-Lactams</b>	
<b>Ram Naresh Yadav<sup>1</sup> and Bimal Krishna Banik<sup>2*</sup></b>	
<sup>1</sup> Department of Chemistry, Faculty of Engineering & Technology, Veer Bahadur Singh Purvanchal University, Jaunpur-222003 (U.P.) India; <sup>2</sup> Department of Mathematics and Natural Sciences, College of Sciences and Human Studies, Deanship of Research, Prince Mohammad Bin Fahd University, Al Khobar 31952, Kingdom of Saudi Arabia; Email: <a href="mailto:bimalbanik10@gmail.com">bimalbanik10@gmail.com</a> ; <a href="mailto:bbanik@pmu.edu.sa">bbanik@pmu.edu.sa</a>	
The reaction of bismuth nitrate and ceric ammonium nitrate on N-para anisyl beta lactam follows two different routes.	

**Design and *in silico* evaluation of new azo barbituric acid analogs as possible anticancer agents**

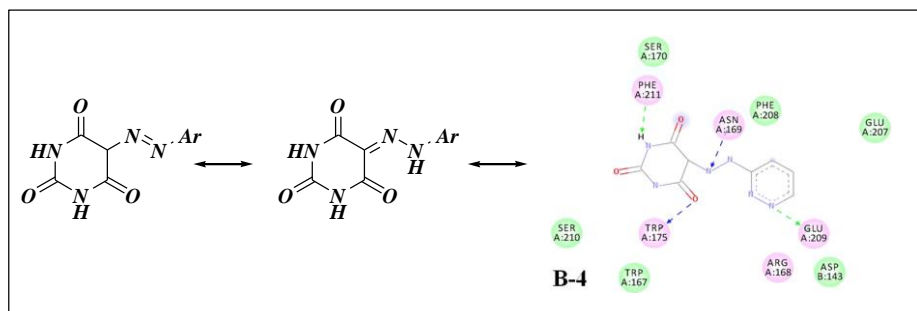
**J. Shaikh<sup>a,b</sup>, A. Pangal<sup>a,b</sup>, I. Tamboli<sup>a</sup>, G. Pote<sup>a</sup> and K. Ahmed<sup>a,b,\*</sup>**

<sup>a</sup>Post-Graduate Department of Chemistry and Research Centre, Abeda Inamdar Senior College of Arts, Science and Commerce (Autonomous), Pune – 411001, India.

<sup>b</sup>Advanced Scientific Research Laboratory, Azam Campus, Pune – 411001, India.

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In the present study, the designed new azo barbituric acid analogs, possible anticancer agents were subjected to *in silico* screening. These compounds were studied for ADMET properties, drug-likeness and toxicity using pKCSM, SwissADME, ProTox-II web servers and followed by molecular docking with caspase-3 protein. The compounds showed good results and devoid of any mutagenicity and immunotoxicity. Using molecular docking studies, it was found that all the compounds interact with caspase-3. The designed compounds thus could be possible anticancer candidates.



**Synthetic Development and Antioxidant Potential of 4-Methyl-8-(methylthio)-2, 6-dioxo 2, 6-dihydro-1H-pyrimido [1,2-a] pyrimidine-7-carbonitrile and its analogues.**

**S. G. Sontakke<sup>a</sup>, D.B. Kadam<sup>b</sup>, S. P. Vartale<sup>c,\*</sup>**

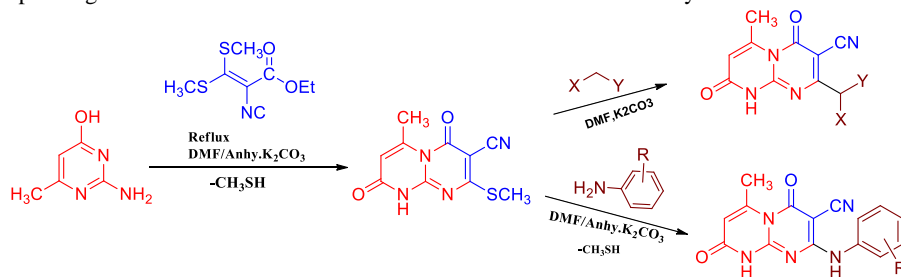
<sup>a</sup>Lonavala Education Trust's Dr. B.N. Purandare Arts, Smt. S.G. Gupta Commerce and Smt. Shardaben Amrutlal Mithaiwala Science College, Lonavala-410403 (M.S), India

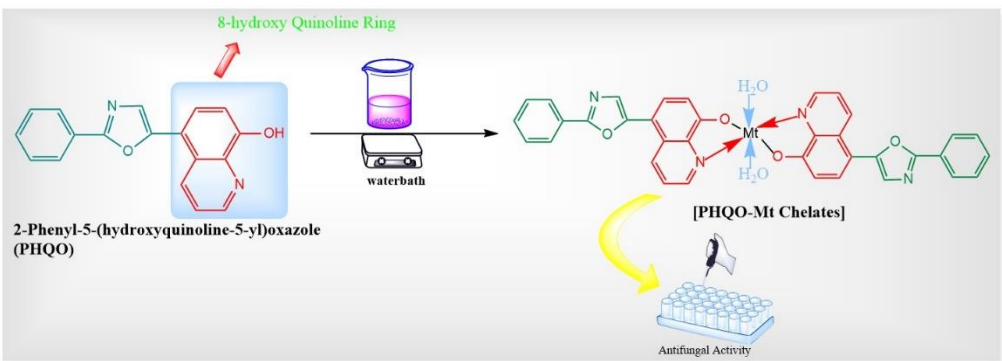
<sup>b</sup>Indira Gandhi (Sr) College, CIDCO, Nanded-431605, (M.S), India

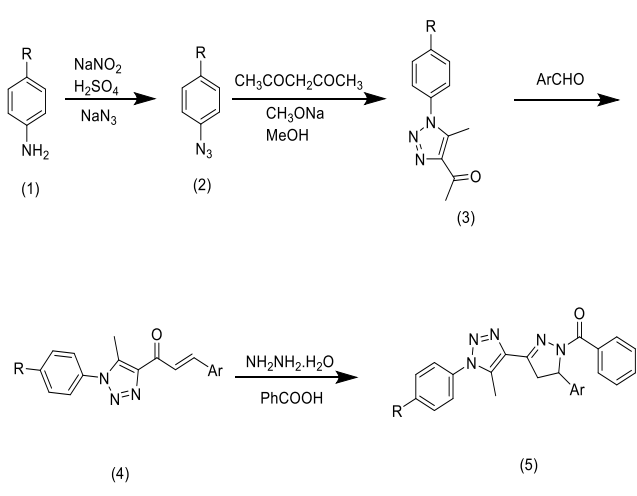
<sup>c</sup>P.G. Research Centre, Dept. of Chemistry, Yeshwant Mahavidyalaya, Nanded-431602, (M.S), India,

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One pot synthesis of 4-methyl-8-(methylthio)-2,6-dioxo-2,6-dihydro-1H-pyrimido[1,2-a]pyrimidine-7-carbonitrile as a parent compound then reacted with some carbon and nitrogen nucleophiles such as active methylene compounds and substituted anilines to afford the corresponding 8-substituted derivatives and studied their antioxidant activity.



Paper-8	Heterocyclic Letters 12: iss.-4 (2022), 745-751
<p><b>Synthesis, Characterization and Antifungal Activity of Metal Chelates of 2-phenyl-5-(8-hydroxyquinolin-5-yl)oxazole</b></p> <p><b>Harsh Patel<sup>*</sup>, Piyush Vyas, Yadav Nishant, Parmar Nayanaben</b></p> <p><i>Department of Chemistry, Sheth M.N. Science College, Patan – 384265, Gujarat, India</i>  <sup>*</sup>Corresponding author Email:</p> <p>2-phenyl-5-(8-hydroxyquinolin-5-yl)oxazole (PHQO) was prepared by condensation of 5-chloroacetyl-8-hydroxy quinoline and benzyl amine. The various transition metal chelates of PHQO with <math>\text{Co}^{2+}</math>, <math>\text{Zn}^{2+}</math>, <math>\text{Ni}^{2+}</math>, <math>\text{Cu}^{2+}</math>, <math>\text{Mn}^{2+}</math> metal ions were prepared and characterized. The antifungal activity of all these derivatives was evaluated against various fungi.</p> 	

Paper-9	Heterocyclic Letters 12: iss.-4 (2022), 753-758
<p><b>Synthesis of (3-(5-methyl-1-phenyl)-1H-1,2,3-triazol-4-yl)-4,5-dihydro-1H-pyrazol-1-yl) phenyl Methanone and its derivatives from substituted Anilines</b></p> <p><b>Ratnamala Sonawane<sup>a</sup>, Mohan Sagare<sup>b</sup></b></p> <p><i><sup>a, b</sup> Department of Chemistry, Institute of science, Homi Bhabha State University, Mumbai-32, Maharashtra, India. Email: <a href="mailto:sagaremohan@yahoo.in">sagaremohan@yahoo.in</a></i></p> 	



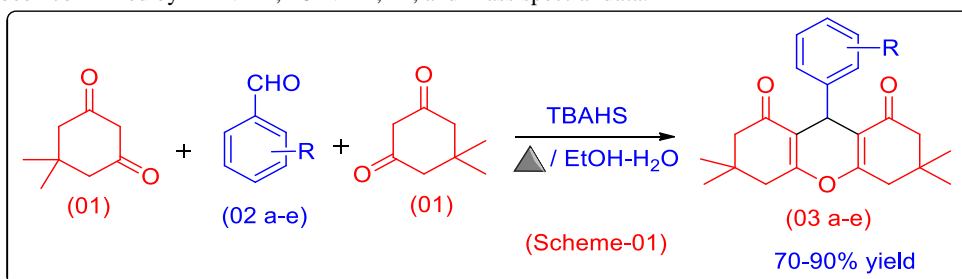
Paper-10

Heterocyclic Letters 12: iss.-4 (2022), 759-765

## Synthesis and Biological activity of tetramethyl-hexahydro-1H-xanthen-1,8(2H)-dione Derivatives

Sunil S. Choudhare<sup>1</sup>, Gopinath S. Khansole<sup>2</sup>, Ashok R.Karad<sup>3</sup>, Navanand B. Wadwale<sup>4</sup>, Vijay N. Bhosale<sup>5</sup><sup>1</sup>Department of Chemistry, S. D. Mahavidyalaya, Soegaon, Aurangabad, (MS), India<sup>2</sup>Department of Chemistry, D. A. B. N. College, Chikhali, Sangli, (MS), India.<sup>3</sup>Department of Chemistry, M. G. M. Ahmedpur, Latur, (MS), India<sup>4</sup>Department of Chemistry, M.S.G College Malegaon, Nashik, (MS), India<sup>5</sup>Department of Chemistry, Yeshwant Mahavidyalaya, Nanded, (MS), India.E-mail: [sunilchoudhare@gmail.com](mailto:sunilchoudhare@gmail.com) Tel.: +91-2438-234395

Tetramethyl-hexahydro-1H-xanthen-1,8(2H)-dione Derivatives (3a-e) have been synthesized by the condensation of two moles of Dimedone and substituted Aromatic aldehydes in ethanol-water mixture using Tetra Butyl Ammonium Hydrogen Sulphate (TBAHS) as a green catalyst. The products have been assayed for their antimicrobial screening against Gram+ve and Gram-ve bacteria. Some of the products showed moderate activity when compared with known standard drug viz. penicillin at the same concentration 10µgm/ml. Spectroscopic techniques are very good tools for the identification of compounds. The structures have been confirmed by <sup>1</sup>H NMR, <sup>13</sup>C-NMR, IR, and Mass spectral data.

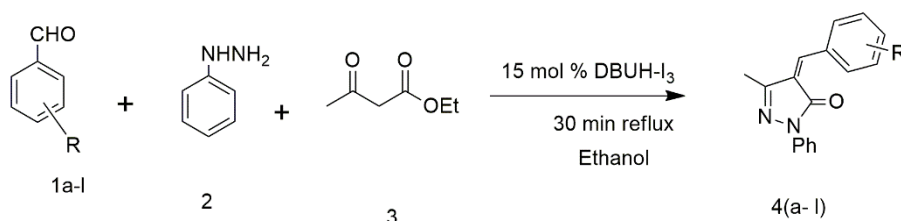


Paper-11

Heterocyclic Letters 12: iss.-4 (2022), 767-773

DBUH-I<sub>3</sub> complex catalysed synthesis of arylidene derivatives of pyrazoleR. Gawade,<sup>a,b</sup> P. Jadhav,<sup>b</sup> S. Shinde,<sup>d</sup> and P. Kulkarni<sup>\*a,c</sup><sup>a</sup>Baburaoji Gholap College Sangvi, Haveli, Pune 411027, Maharashtra, India<sup>b</sup>S. M. Joshi College Hadapsar, Haveli, Pune 411028, Maharashtra, India<sup>c</sup>Post Graduate Research Center in Organic Chemistry and Department of Chemistry Hutatma Rajguru Mahavidyalaya, Rajgurunagar Pune 410505, Maharashtra, India<sup>d</sup>Annasaheb Awate College Manchar, Pune 410503, Maharashtra, India

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**Green synthesis, characterization and antimicrobial evaluation of some piperazine benzenamine imino derivatives**

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<sup>1</sup>Department of Chemistry, K. R. Mangalam University, Gurugram, 122103, Haryana, India

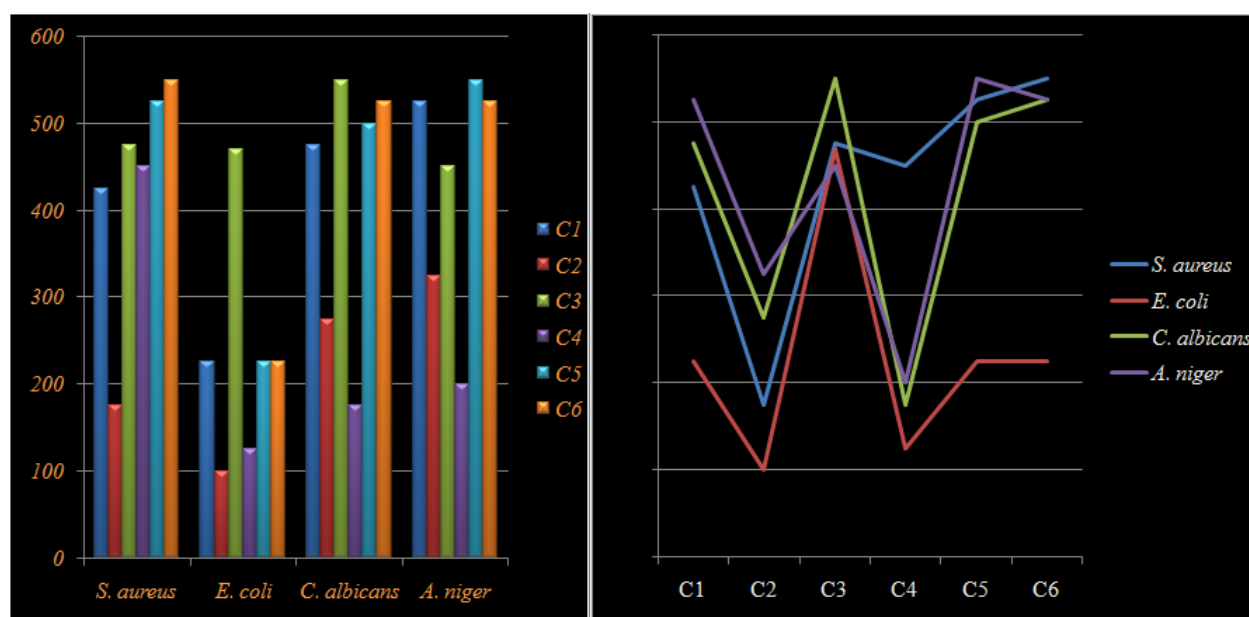
<sup>2</sup>Therachem Research Medilab, Sitapura, Jaipur, 302022, Rajasthan, India

<sup>3</sup>Department of Chemistry, Jaipur National University, Jaipur, 302017, Rajasthan, India

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The interaction of piperazine benzenamine (**A**) and substituted benzaldehyde (**B**) via a greener approach resulted in the synthesis of their imino derivatives (**C1 – C6**). The compound **A** was synthesized by green method using water as a solvent. The resultant compounds were characterized by elemental analysis, <sup>1</sup>H-NMR, IR spectral studies.

The *in-vitro* Antimicrobial activities of all the final compounds have proven that they are active against bacterial strains *S. aureus*, *E. coli* and fungal strains *C. albicans*, *A. niger*.



**Figure 1(a & b).** Antimicrobial activities of synthesized compounds C1-C6 (Y-axis stands for MIC data in µg/ml)



### Microwave synthesis and evaluation of antimicrobial activity of novel 3,5 disubstituted-2-pyrazolines

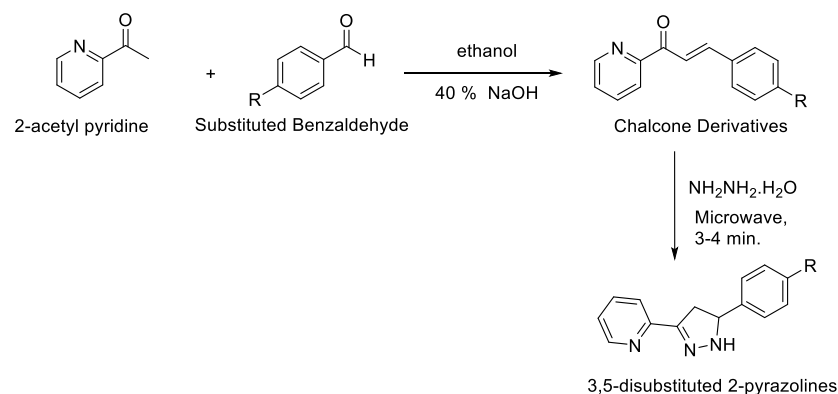
S.L. Kumbhare<sup>a\*</sup>, Y.K. Meshram<sup>b</sup>

<sup>a</sup>Department of Chemistry, Shri Shivaji Science & Arts College, Chikhli, Dist.- Buldana (M.S.), 443201, India

<sup>b</sup>Department of Chemistry, G.S. College, Khamgaon, Dist.- Buldana (M.S.), 444303, India

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The rapid, simple microwave-assisted synthesis of 3(pyridine-2-yl)-5-Phenyl-2-Pyrazoline derivatives. Reactions are performed neat at 600 watt for 2-3 minutes.



### One-pot three component synthesis of benzylidene conjugated-1, 2, 4-triazines in glycerol

V. Anitha Rani<sup>\*1</sup> and Y. Bharathi Kumari<sup>2</sup>

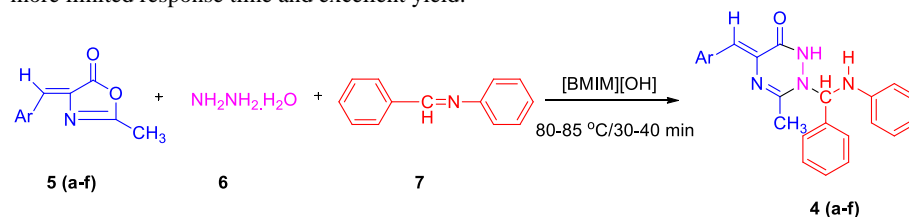
1. Department of Chemistry, Institute of Aeronautical Engineering, Dundigal, Hyderabad

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College of Engineering, Kukatpally, Hyderabad (A.P), India - 500 085.

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One-pot three component reaction for the synthesis of Benzylidene conjugated-1, 2, 4-triazines **4a-4f** were depicted by the combination of (Z)- 4-(benzylidene)-2-methyl-oxazol-5(4H)- ones **1a-1f** with hydrazine hydrate **2** followed by Schiff base **3** in Glycerol as solvent in the presence of DBU as base catalyst for 30-40 min at 80-85 °C. The significance of this reaction incorporates more limited response time and excellent yield.



**Synthesis, molecular docking, molecular properties estimations, and antiinflammatory activity of 5, 7-dihydroxy-3'-prenyl flavanone**

**Badikela Rama Krishna<sup>a</sup>, Rathnakar Ch<sup>a</sup>, Durgaprasad Kemiseti<sup>b</sup>, Vijayaraj Surendran<sup>c</sup> and Parameshwar Ravula<sup>c,\*</sup>**

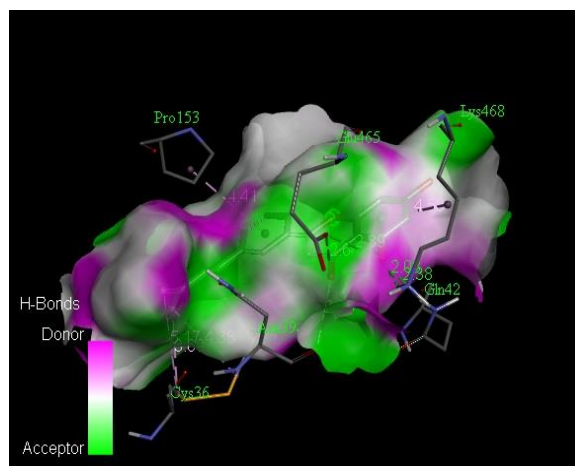
<sup>a</sup>Department of Pharmaceutical Analysis & Quality Assurance, Guru Nanak Institutions Technical Campus-School of Pharmacy, Hyderabad, Telangana, India

<sup>b</sup>Faculty of Pharmaceutical Science, Assam Down Town University, Guwahati, Assam, India

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A novel compound of 5, 7-dihydroxy-3'-prenyl flavanone was synthesized by facile synthetic method and characterized by physical and spectral data. The anti-inflammatory screening of the synthesized compound was performed by *in vivo* using carragenan induced paw oedema method. *In silico* prediction of molecular and drug-likeness properties of synthesized compound was promising. Moreover, the synthesized compound showed significant docking interactions with COX-2 active site. Molecular docking results along with the biological data suggested that the tested compounds have the potential as valuable lead for antiinflammatory activity.

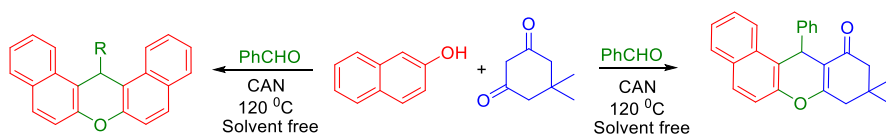


**CAN mediated Multicomponent synthesis of benzoxanthene and benzochromene Libraries**

**Sneha Kumari<sup>1</sup>, Sanjeev Kumar Jha<sup>\*</sup>**

<sup>1</sup>Department of Chemistry, M. L. T. College Saharsa (A Constituent unit of B. N. Mandal University Madhepura), Bihar, India

<sup>\*</sup>Department of Chemistry, M. L. T. College Saharsa (A Constituent unit of B. N. Mandal University Madhepura), Bihar, India



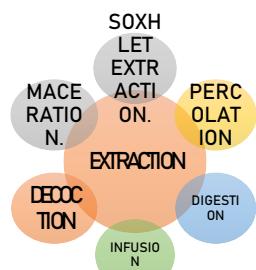




Paper-17	Heterocyclic Letters 12: iss.-4 (2022), 821-829
<p><b>Synthesis, characterization, antimicrobial activity and dna cleavage study of (E)-3-(((3-hydroxypyridin-2-yl) imino) methyl) quinolin-2-ol schiff base metal complexes</b></p> <p><b>Anilkumar Ambala<sup>a&amp;c</sup>, D. Ravikumar<sup>b</sup>, Balraj. G<sup>c</sup>, K. Santhosh Kumar<sup>a</sup> and Ch. Abraham Lincoln<sup>c*</sup></b></p> <p><sup>a</sup>Department of Freshman Engineering, Geethanjali College of Engineering and Technology, Cheeryala, Telangana, India-501301.  <sup>b</sup>Methodist college of engineering and technology, Abids, Hyderabad  <sup>c</sup>Department of chemistry, Osmania University, Hyderabad-500 007, India.                  Email ID: <a href="mailto:lincoln.ab86@gmail.com">lincoln.ab86@gmail.com</a>.</p> <p>A series of novel (E)-3-(((3-hydroxypyridin-2-yl)imino)methyl)quinolin-2-ol based metal complexes of Copper (II), Cobalt (II), Nickel (II), Zinc (II) and Manganese (II) were synthesized. The ligand was synthesized by the condensation of 2-Hydroxyquinoline-3-carbaldehyde and 2-aminopyridin-3-ol in alcoholic medium and the metal complexes were synthesized by 1:1 metal to ligand ratio. The ligand and metal complexes were characterized by different analytical techniques such as FT-IR, UV-visible, Mass spectrometry, SEM, EDX, TGA and magnetic moment measurements. The ligand and all the metal complexes were tested for their antimicrobial activity and DNA cleavage studies. The antimicrobial screening results suggested that most of the metal complexes showed better activity compared to the free Schiff base ligand. In the DNA cleavage studies pUC18 DNA is used for cleavage experiment, complete cleavage of DNA occurred with the Nickel and Manganese complexes and Partial cleavage of DNA occurred with Zinc and copper complexes.</p> <p style="text-align: center;"><b>M(II) Chloride</b>  <b>Methanol</b>  <b>M=Cu,Co,Ni, Zn and Mn</b></p>	

Paper-18	Heterocyclic Letters 12: iss.-4 (2022), 831-843
<p><b>Synthesis, biological evaluation and insilico studies of thieno[2,3-d]pyrimidine-1,2,3-triazoles</b></p> <p><b>Begari Nagaraju<sup>a</sup>, Vulichi R Srinivasa<sup>b</sup>, Settyapalli Triloknadh<sup>a</sup>, Maddineni Aruna Kumari<sup>a</sup>, Muthirevula Rajeswari<sup>a</sup>, Chunduri Venkata Rao<sup>a*</sup></b></p> <p><sup>a</sup>Department of Chemistry, <sup>b</sup>department of Biochemistry, Sri Venkateswara University, Tirupati 517 502, Andhra Pradesh, India                  Tel: +91-9849605140, E-mail: <a href="mailto:cvrsvu@gmail.com">cvrsvu@gmail.com</a></p> <p style="text-align: center;"><b>6a-j</b>                  Yield 80-92 %</p>	

## REVIEWS

Review no.1	Heterocyclic Letters 12: iss.-4 (2022), 845-853
<b>Review on methods used in isolations of phytochemicals from medicinal plants</b>	
<b>Ajay B. Jadhao</b>	
<p>Department of Botany, Arts and Science College Pulgaon, District Wardha                  Affiliated by RTM Nagpur University                  Email Id- <a href="mailto:cyrusajay@gmail.com">cyrusajay@gmail.com</a></p>	
<p>Plants contain secondary metabolites such as Alkaloids, flavonoids, phenolics, tannins, saponins, steroids, glycosides, terpenes, etc. Today, it is very crucial to develop effective and selective methods for the extraction and isolation of new natural products. Phytochemicals are now determined using a variety of contemporary techniques maceration, infusion, decoction, percolation, digestion, Soxhlet extraction, aqueous-alcoholic extraction by fermentation, supercritical fluid extraction, etc. These types of techniques are very useful in the extraction process. however, qualitative assays are still widely used for basic phytochemical screening of plants. So, for convenience, for newly researcher, the review is focusing on some important methods used to isolate phytochemicals present in plants.</p>	
	

Review no.2	Heterocyclic Letters 12: iss.-4 (2022), 855-869
<b>Review: Chalcones as natural products and their derivatives in biological activities</b>	
<b>F.M.Zahou<sup>a</sup>, Ruba A. Alolayan<sup>b</sup>, Nadia A.A.Elkanzi<sup>b*</sup></b>	
<p><sup>a</sup>Biology Department, college of Science, Jouf University, sakaka, 2014, Saudi Arabia  <sup>b</sup>Chemistry Department, college of Science, Jouf University, sakaka, 2014, Saudi Arabia                  Corresponding author (N.A.A.Elkanzi) &amp; E-mail: <a href="mailto:nahasan@ju.edu.sa">nahasan@ju.edu.sa</a> &amp; <a href="mailto:kanzi20@yahoo.com">kanzi20@yahoo.com</a></p>	
<p>This review provides pharmacological activity, synthesis and biological activity of chalcones</p>	
